Atmospheric chemistry and aerosols (II)

Dust in the Martian atmosphere



Martian dust storms span the entire planet, in June 2018. The image was taken from the NASA's rover *Curiosity*

Dust in the Martian atmosphere



- Micrometer-sized small mineral particles float in the atmosphere with a background optical thickness of 0.1-0.5
- The dust loading changes with time and space.
- The dust serves as a heat source in the atmosphere by absorbing sunlight.

Seasonal variation of optical thickness in infrared (Smith et al. 2004)

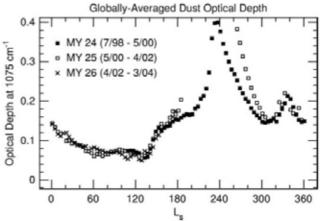


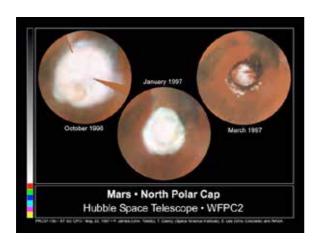
Fig. 7. Globally-averaged daytime (local time \sim 1400) dust optical depth at 1075 cm⁻¹ (scaled to an equivalent 6.1-mbar pressure surface) as a function of season (L_s). Three martian years are represented: Mars Year 24 (MY 24) (\blacksquare), MY 25 (\square), MY 26 (\times). During the planet-encircling dust storm of 2001 (MY 25), globally-averaged dust opacity reached 1.3 at $L_s = 205-215^\circ$.

Seasons of Mars

Ls = 0 : spring equinox of the northern hemisphere

Ls = 270

Ls = 180

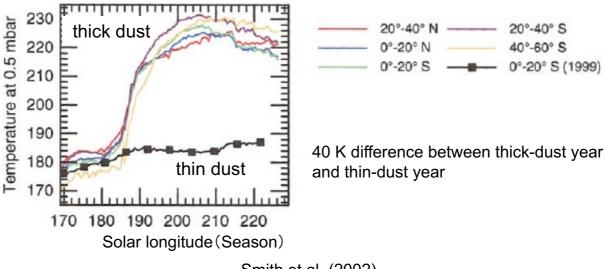


1.66 AU

- 火星は公転軌道の離心率が大きいため 季節変化が著しく南北非対称
- 南半球の夏に太陽までの距離が近くなる

Dust as a heat source

- Absorption of solar radiation
 - much stronger than the greenhouse effect of CO₂, which is only several kelvins
 - much stronger than cloud albedo effect and latent heat



Smith et al. (2002)

Radiative-convective model

Gierasch & Goody (1972)

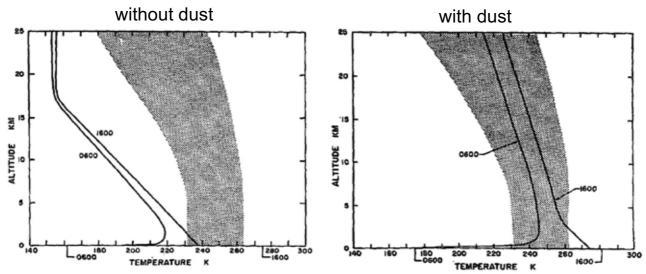


Fig. 1. Martian temperature calculations. The stippled area represents temperatures reported by Kliore et al. (1972) and Hanel et al. (1972). The lines are theoretical profiles for a pure CO₂ atmosphere, at 1600 and at 0600 hours (the coldest time). Both theory and observation refer to mid-latitude summer conditions. The tags indicate the ground temperatures. In the case of the 1600 theoretical profile a strong boundary layer is indicated.

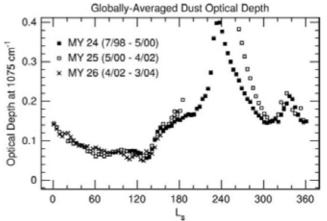
Fig. 2. Same as Fig. 1 except that the atmosphere contains an extra solar absorber, evenly mixed with the atmosphere at all levels, and having an optical depth of 0.10 at all wavelengths. Note the weak boundary layer at 1600.

Dust storms on Mars



regional storm

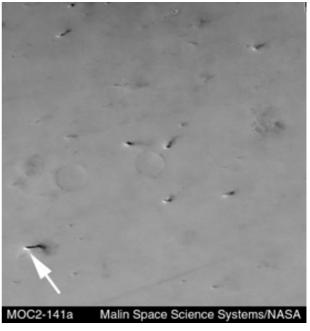
Seasonal variation of optical thickness in infrared (Smith et al. 2004)





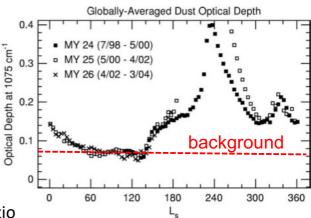
Dust devils

Source of background atmospheric dust?

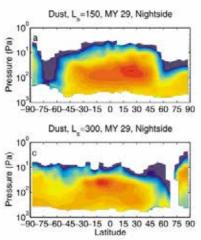


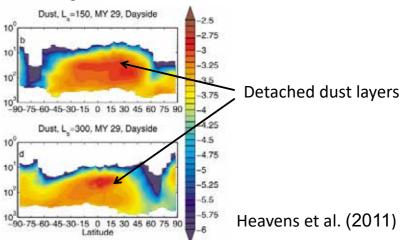
Distribution of atmospheric dust

- Origin of the "background" dust is unknown
- Maximum mixing ratio at 10–20 km altitudes



Meridional distribution of dust mixing ratio





DESCRIPTION OF STREET OF STREET

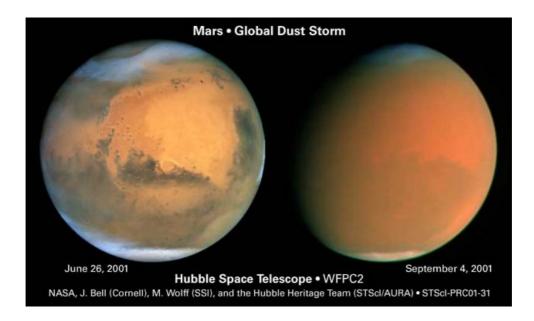
"Rocket dust storm"

Modeling by Spiga et al. (2013)

Dust plumes continuously get buoyancy through solar heating

Figure 12. The LMD-MMM storm simulation with lifting and no initial dust perturbation. Same as Figure 4 except that local times range from 0800 to 1800 and longitude-altitude sections are obtained at latitude 1.5°S.

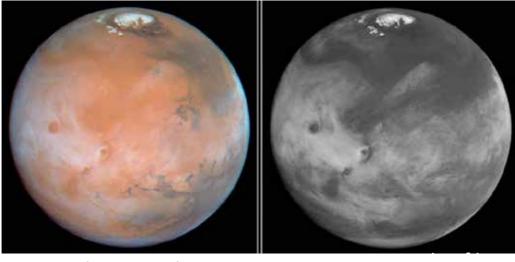
Global dust storm



- · Global dust storms tend to occur in southern spring-summer
- Positive feedback between dust heating and the intensification of winds is expected in the development of global dust storms.

H₂O ice clouds on Mars

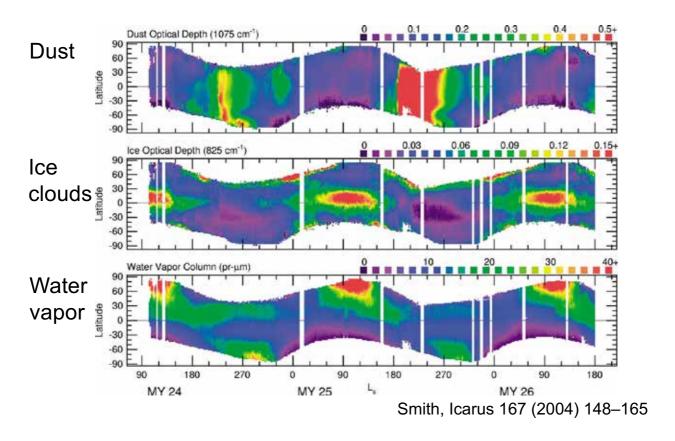
HST Mars image



color composite

blue (410 nm)

Seasonal variation of dust, clouds, and H₂O vapor observed by an infrared spectrometer (TES) on Mars Global Surveyor



Seasonal cycle of Martian water

- 北極冠の消長が全体を駆動
- 北半球の春~夏に北極冠が昇華して北極域の水蒸気濃度が上昇、これが(この時期 の弱い)水平渦で低緯度に拡散的に運ばれる。
- 低緯度に運ばれた水蒸気の一部は赤道越えのハドレー循環で南半球へ
- ・ 北半球の秋~冬には北極冠で 凝結により水蒸気濃度が低下 し、南北濃度勾配が逆転する ため、傾圧不安定などに伴う水 平渦で低緯度から北極域に水 蒸気が拡散的に戻る。低緯度 の水蒸気量はそれまでの水蒸 気輸送の履歴で決まる。

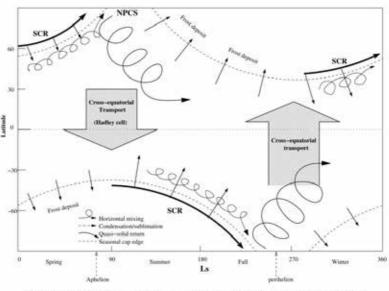


Figure 3. Chart describing the principal events affecting the Martian water cycle over the course of a year. NPCS stands for North Polar Cap Sublimation; SCR stands for Seasonal Cap Recession.

Polar caps: H₂O ice + CO₂ ice

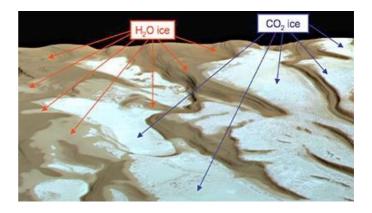
- Seasonal variation
- · Residual polar caps in summer
 - H₂O only on the north
 - H₂O + CO₂ on the south ← mystery
- Southern CO₂ ice seems to serve as a cold trap of H₂O

North



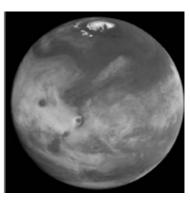
South

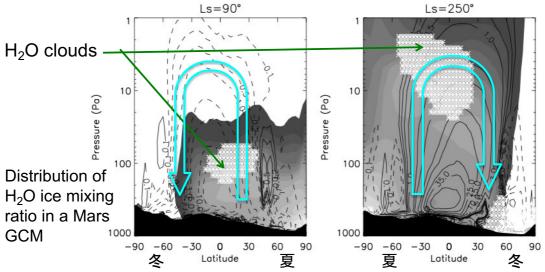




Water transport by Hadley circulation

 Warmer southern summer than northern favors net northward transport of water.

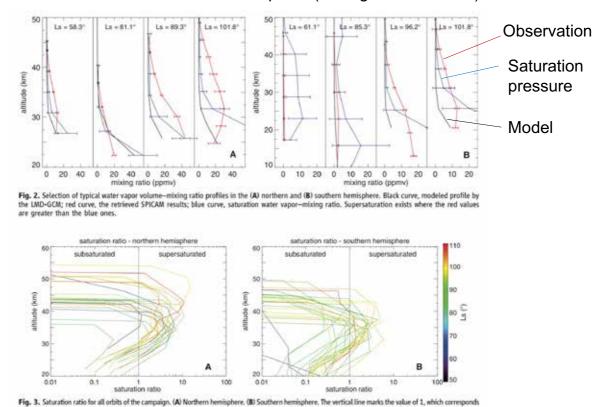




Montmessin et al. (2004)

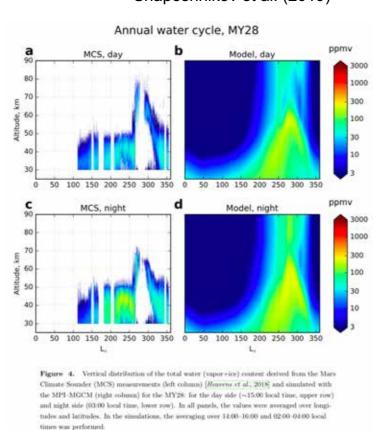
Supersaturation of water vapor on Mars

SPICAM on Mars Express (Maltagliati et al. 2011)



Vertical distribution of water vapor on Mars during the course of a Mars year Shaposhnikov et al. (2019)

to the saturated state

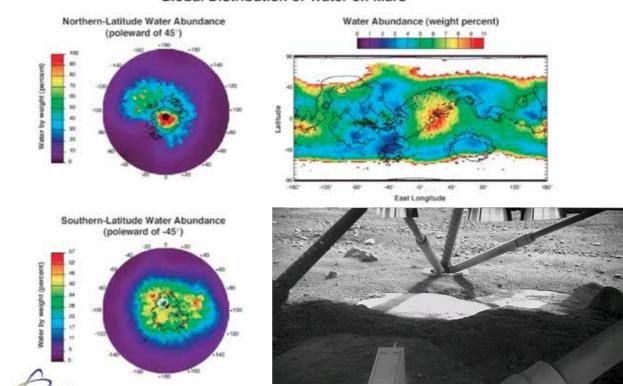


Mars Odyssey

Neutron Spectrometer (NS) and High-Energy Neutron Detector (HĔND)

Subsurface ice





Phoenix Mars Lander

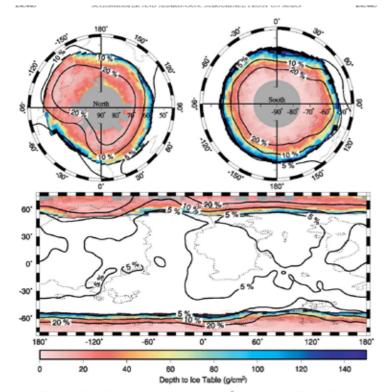


Figure 8. Color indicates depth to the ice table in g cm⁻² when ice is in equilibrium Figure 8. Color indicates depth to the ice table in g cm⁻² when ice is in equilibrium with the atmospheric water vapor. Goodand ice is unstable in the white area. Black segments indicate finite barial depths larger than 150 g cm⁻². Missing data points are shown in gray. Assumed volume fraction of ice is 40%, but the geographic boundary between icy and ice-free soil is independent of the ice fraction. Solid contours indicate water-equivalent hydrogen content in percent determined from neuron spectroscopy [Feldman et al., 2004]. The dotted lines are 200 J m⁻²K^{-1/2} contours of thermal inertia.

Schorghofer and Aharonson (2005)

Comparison with models

Near equilibrium?

色: 大気中の水蒸気との平衡 状態を仮定して計算される氷床 までの深さ

実線: 中性子分光観測から見 積もられた水含有量

→現在の気候とだいたい 平衡状態か

Chemistry of gas giants

Lodders, 2010

Many of the gases observed in their atmospheres are hydrides, which are thermodynamically stable forms in the H_2 -rich atmospheres (e.g., CH_4 , NH_3 , H_2O , H_2S , PH_3 , GeH_4 , and AsH_3).

These gases (except H_2O and H_2S) are photochemically destroyed by UV sunlight in the stratosphere to produce disequilibrium species (e.g., ethane C_2H_6 , acetylene C_2H_2 , and ethylene C_2H_4 , hydrazine N_2H_4).

The disequilibrium species react with H₂ to reform hydrides once they are transported downward into the hot, high pressure regions.

Gas	Jupiter ^a	Saturn	Uranus	Neptune
H ₂	$86.4 \pm 0.3\%$	$88 \pm 2\%$	~82.5 ± 3.3%	~80 ± 3.2 %
⁴ He	$13.6 \pm 0.3\%$	$12 \pm 2\%$	15.2 ± 3.3 %	19.0 ± 3.2 %
CH4	$(1.81 \pm 0.34) \times 10^{-3}$	$(4.7 \pm 0.2) \times 10^{-3}$	~2.3 %	~1-2 %
NH_3	$(6.1 \pm 2.8) \times 10^{-4}$	$(1.6 \pm 1.1) \times 10^{-4}$	<100 ppb	<600 ppb
H ₂ O	520 ⁺³⁴⁰ ₋₂₄₀ ppm	2-20 ppb		
H ₂ S	$67 \pm 4 \text{ ppm}$	<0.4 ppm	<0.8 ppm	<3 ppm
HD	$45 \pm 12 \text{ ppm}$	$110 \pm 58 \text{ ppm}$	~148 ppm	~192 ppm
¹³ CH ₄	19 ± 1 ppm	51±2 ppm		
C ₂ H ₆	$5.8 \pm 1.5 \text{ ppm}$	$7.0 \pm 1.5 \text{ ppm}$		
PH ₃	$1.1 \pm 0.4 \text{ ppm}$	$4.5 \pm 1.4 \text{ ppm}$		
CH ₃ D	$0.20 \pm 0.04 \text{ ppm}$	$0.30 \pm 0.02 \text{ ppm}$	~8.3 ppm	~12 ppm
C ₂ H ₂	$0.11 \pm 0.03 \text{ ppm}$	$0.30 \pm 0.10 \text{ ppm}$	~10 ppb	60 ⁺¹⁴⁰ ₋₄₀ ppb
HCN	$60 \pm 10 \text{ ppb}$	<4 ppb	<15 ppb	$0.3 \pm 0.15 \text{ ppb}$
HC ₃ N			<0.8 ppb	<0.4 ppb
C ₂ H ₄	7 ± 3 ppb	~0.2 ppbb		
CO ₂	5-35 ppb	0.3 ppb	40 ± 5 ppt	
C ₂ H ₆			10 ± 1 ppb	1.5 ^{+2.5} _{-0.5} ppm
CH ₃ C ₂ H	2.5 ⁺² ₋₁ ppb	0.6 ppb	0.25 ± 0.03 ppb	
CO	$1.6 \pm 0.3 \text{ ppb}$	$1.4 \pm 0.7 \text{ ppb}$	<40 ppb	0.65 ± 0.35 ppm
CH ₃ CN				<5 ppb
GeH ₄	0.7 ^{+0.4} _{-0.2} ppb	$0.4 \pm 0.4 \text{ ppb}$		
C ₄ H ₂	$0.3 \pm 0.2 \text{ ppb}$	0.09 ppb	0.16 ± 0.02 ppb	
AsH ₃	$0.22 \pm 0.11 \text{ ppb}$	$2.1 \pm 1.3 \text{ ppb}$		

assuming a total stratospheric column density of 1.54%1025 cm-2.

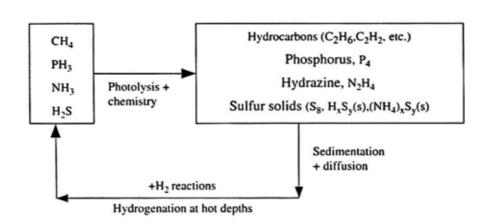
From Lodders & Fegley 1998 and updates: Mahaffy et al. 2000, Atreya et al. 2003, Lodders 2004, Wong et al. 2004

Cycle of hydrogen-bearing species on giant planets

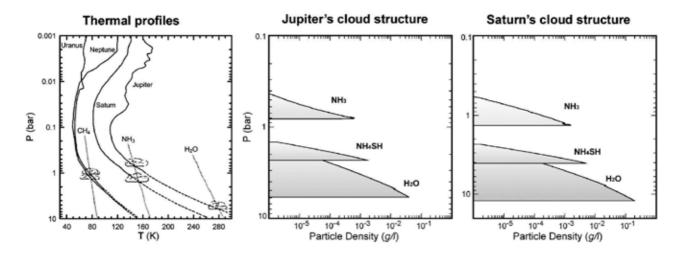
$$2CH_4 + photons \rightarrow \underbrace{C_2H_6 + H_2}_{ethane + hydrogen}$$

$$2nCH_4 + photons \rightarrow \underbrace{(C_2H_2)_n + 3nH_2}_{polyacetylene + hydrogen}$$

$$6nCH_4 + photons \rightarrow \underbrace{(C_6H_6)_n + 9nH_2}_{polyaromatic hydrocarbons + hydrogen}$$
Catling & Kasting (2017)



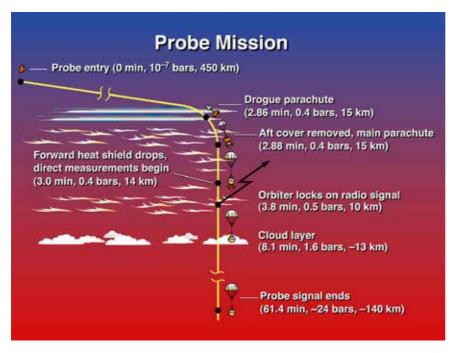
Clouds



Sanchez-Lavega et al.



Galileo probe (entry: December 7, 1995)





Dry atmosphere?

- Brightness of the sky abruptly drops off at a pressure level of 0.6 bars, indicating an ammonia cloud layer above this height. The tenuous cloud layer detected by the NEP was not seen by this experiment.
- Clouds are patchy and that the Probe went through a relatively clear area.
 - Pressure Altitude (bars) (km)

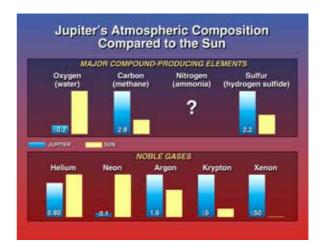
 0.5 +18
 Ammonia ke (loud?

 2 -21
 Distributed solar heating

 -10 Watts/meter 0 0 Watts/meter 8
 Upward Net Flux

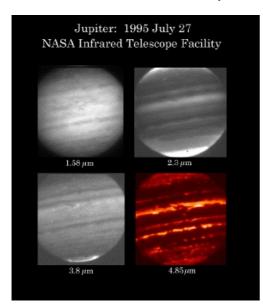
 Upward Net Flux

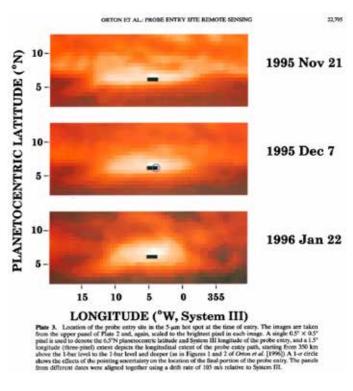
- The atmosphere has much less oxygen than the Sun's atmosphere, implying a surprisingly dry atmosphere.
- Planetary scientists had expected oxygen to be enriched relative to the solar value due to impacts by comets and other small bodies over the 4.5 billion years.



The probe apparently entered a special location

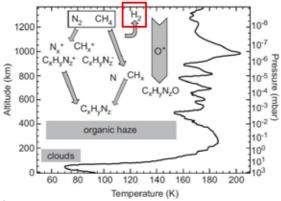
The Probe entry site is near the edge of a so-called infrared "hot spot". These "hot spots" are believed to represent regions of diminished clouds on Jupiter.





Orton et al. 1998

Atmospheric chemistry on Titan

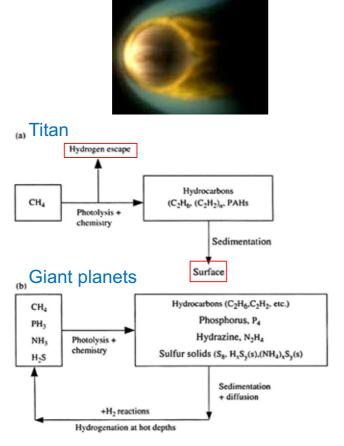


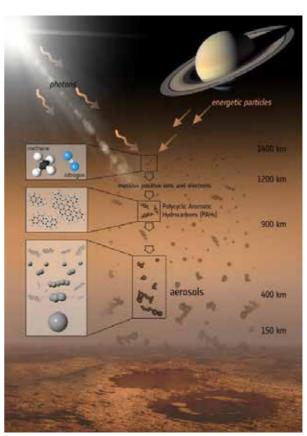
Vuitton et al. (2014)

Figure 7.1 Summary of our current understanding of Titan's atmospheric chemistry, from N_2 and CH_4 to minor gas species, then to macromolecules and organic aerosols. The temperature profile is a smoothed version of the HASI profile (Fulchignoni et al., 2005).

Atmospheric composition of Titan (Coustenis 2007)

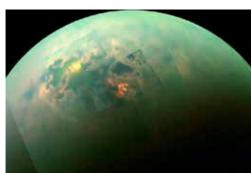
Constituent	Mole Fraction (atm. altitude level)			
Major				
Molecular nitrogen, N ₂	0.98			
Methane, CH ₄	4.9×10^{-2} (surface)			
	$1.41.6\times10^{-2}(\text{stratosphere})$			
Monodeuterated methane, CH ₃ D	$6\times10^{-6}(\text{in CH}_3\text{D, in stratosphere.})$			
Argon, 36Ar	2.8×10^{-7}			
40 _{Ar}	4.3×10^{-5}			
Minor				
Hydrogen, H ₂	~0.0011			
Ethane, C ₂ H ₆	1.5×10^{-5} (around 130 km)			
Propane, C ₃ H ₈	5×10^{-7} (around 125 km)			
Acetylene, C ₂ H ₂	4×10^{-6} (around 140 km)			
Ethylene, C ₂ H ₄	1.5×10^{-7} (around 130 km)			
Methylacetylene, CH ₃ C ₂ H	$6.5\times10^{-9}(\text{around }110\text{km})^{\text{s}}$			
Diacetylene, C ₄ H ₂	$1.3\times10^{-9}(\text{around}\ 110\ \text{km})^{\alpha}$			
Cyanogen, C ₂ N ₂	$5.5\times10^{-9}(\text{around }120\text{km})^{\text{s}}$			
Hydrogen cyanide, HCN	$1.0\times10^{-7}(\text{around }120\;\text{km})^{\alpha}$			
	5×10^{-7} (around 200 km) ^b			
	$5\times10^{-6}(\text{around 500 km})^{b}$			
Cyanoacetylene, HC ₃ N	1×10^{-9} (around 120 km) ^a			
	1×10^{-7} (around 500 km) ^b			
Acetonitrile, CH ₃ CN	1×10^{-8} (around 200 km) ^c			
	1×10^{-7} (around 500 km)			
Water, H ₂ O	$8 \times 10^{-9} (at 400 \text{ km})^d$			
Carbon monoxide, CO	4×10^{-5} (uniform profile) ^c			
Carbon dioxide, CO2	1.5 × 10 ⁻⁸ (around 120 km)			



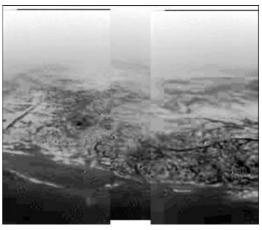


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Cassini's Visual and Infrared Mapping Spectrometer (VIMS)



Huygens' touchdown

