# Atmospheric chemistry and aerosols (II)

### Dust in the Martian atmosphere



Martian dust storms span the entire planet, in June 2018. The image was taken from the NASA's rover *Curiosity* 

### Dust in the Martian atmosphere



- Micrometer-sized small mineral particles float in the atmosphere with a background optical thickness of 0.1-0.5.
- The dust loading changes with time and space.
- The dust serves as a heat source in the atmosphere by absorbing sunlight.

Seasonal variation of optical thickness in infrared (Smith et al. 2004)



Fig. 7. Globally-averaged daytime (local time ~ 1400) dust optical depth at 1075 cm<sup>-1</sup> (scaled to an equivalent 6.1-mbar pressure surface) as a function of season ( $L_s$ ). Three martian years are represented: Mars Year 24 (MY 24) (**I**), MY 25 (**I**), MY 26 (×). During the planet-encircling dust storm of 2001 (MY 25), globally-averaged dust opacity reached 1.3 at  $L_s = 205-215^{\circ}$ .



- The north-south asymmetry of the seasonal cycle is large due to the large orbital eccentricity.
- The distance to the sun gets closer in the southern summer.

### Dust as a heat source

- Absorption of solar radiation ٠
  - much stronger than the greenhouse effect of CO<sub>2</sub>, which is only several kelvins
  - much stronger than cloud albedo effect and latent heat



Globally-Averaged Dust Optical Depth

dust year

40 K difference between

thick-dust year and thin-

### Dust storms on Mars



### regional storm

24 (7/98 - 5/00) Optical Depth at 1075 cm<sup>-1</sup> AY 25 (5/00 - 4/02) 0.3 Y 26 (4/02 - 3/04) 0.2 Seasonal variation of optical 0.1 thickness in infrared (Smith et al. 2004) 0 0 60 120 180 240 300 360 L,

0.4



## Dust devils

 Source of background atmospheric dust ?



## Distribution of atmospheric dust

- Origin of the "background" dust is unknown
- Maximum mixing ratio at 10–20 km altitudes

#### Globally-Averaged Dust Optical Depth 0.4 MY 24 (7/98 - 5/00) Optical Depth at 1075 cm<sup>-1</sup> MY 25 (5/00 - 4/02) 0.3 MY 26 (4/02 - 3/04) 0.2 0.1 background 0 60 120 180 240 300 360 0 Ls -2.5 -2.75 -3 3.25

### Meridional distribution of dust mixing ratio





#### Smith et al. (2004)



"Rocket dust storm" Modeling by Spiga et al. (2013)

Dust plumes continuously get buoyancy through solar heating

### Global dust storm



- Global dust storms tend to occur in southern spring-summer
- Positive feedback between dust heating and the intensification of winds is expected in the development of global dust storms.

Figure 12. The LMD-MMM storm simulation with lifting and no initial dust perturbation. Same as Figure 4 except that local times range from 0800 to 1800 and longitude-altitude sections are obtained at latitude 1.5°S.

### H<sub>2</sub>O ice clouds on Mars

### HST Mars image



color composite

blue (410 nm)

Seasonal variation of dust, clouds, and H<sub>2</sub>O vapor observed by an infrared spectrometer (TES) on Mars Global Surveyor



## Seasonal cycle of Martian water

Water cycle is basically driven by the sublimation from/condensation onto the north polar cap.



Figure 3. Chart describing the principal events affecting the Martian water cycle over the course of a year. NPCS stands for North Polar Cap Sublimation; SCR stands for Seasonal Cap Recession.

### Polar caps: $H_2O$ ice + $CO_2$ ice

- Seasonal variation
- Residual polar caps in summer
  - H<sub>2</sub>O only on the north
  - H<sub>2</sub>O + CO<sub>2</sub> on the south
- Southern CO<sub>2</sub> ice seems to serve as a cold trap of H<sub>2</sub>O (Montmessin et al. 2007)

North



South





### Water transport by Hadley circulation





## SPICAM on Mars Express (Maltagliati et al. 2011)



Fig. 2. Selection of typical water vapor volume-mixing ratio profiles in the (A) northern and (B) southern hemisphere. Black curve, modeled profile by the LMD-GCM; red curve, the retrieved SPICAM results; blue curve, saturation water vapor-mixing ratio. Supersaturation exists where the red values are greater than the blue ones.



Fig. 3. Saturation ratio for all orbits of the campaign. (A) Northern hemisphere. (B) Southern hemisphere. The vertical line marks the value of 1, which corresponds to the saturated state.

## Dependence of volatile escape on lower atmosphere processes



Meridional distribution of zonal-mean temperature obtained by MGS/TES (Smith et al. 2001)



## Meridional cross sections in Mars GCM (Shaposhnikov et al. 2019)





### New concept



Martian Moons eXploration (MMX) JAXA's next-generation sample return mission

- Launch in 2024
- Phobos & Deimos
  - Remote sensing & in situ observation (Phobos)
  - Retrieve samples (>10 g) from Phobos & return to Earth in 2029
- Mars: Remote sensing mainly from the Phobos orbit
- First sample return mission from the Martian system

# Instruments for Mars atmosphere observation

### • OROCHI

- Wide-angle camera, 8 colors
- 3 colors (480, 650, 950 nm) will be used for Mars observation.
- 2.5 km/pix (sub S/C) from QSO

### • TENGOO

- Narrow-angle camera
- 35 m/pix (sub S/C) from QSO
- MIRS
  - Push-bloom type spectrometer
  - Spectral resolution: 10nm
  - Spectral bandpass: 0.9–3.6 µm
  - 2.1 km/pix (sub S/C) from QSO





Barucci et al. (2021)

## Continuous global monitoring from Martian orbit



### Mars Odyssey

Neutron Spectrometer (NS) and High-Energy Neutron Detector (HEND)

## Subsurface ice

Global Distribution of Water on Mars



### Chemistry of gas giants



Many of the gases in the gas giants are hydrides, which are thermodynamically stable (e.g.,  $CH_4$ ,  $NH_3$ ,  $H_2O$ ,  $H_2S$ ,  $PH_3$ ,  $GeH_4$ , and  $AsH_3$ ).

These gases (except  $H_2O$  and  $H_2S$ ) are photochemically destroyed by solar UV to produce disequilibrium species (e.g., ethane  $C_2H_6$ , ethylene  $C_2H_4$ , acetylene  $C_2H_2$ , hydrazine  $N_2H_4$ ).

The disequilibrium species react with H<sub>2</sub> to reform hydrides once they are transported downward into the hot, high pressure regions.

| Gas                              | Jupitera                                | Saturn                         | Uranus                      | Neptune                                 |
|----------------------------------|---|--------------------------------|-----------------------------|---|
| H <sub>2</sub>                   | 86.4 ± 0.3%                             | $88 \pm 2\%$                   | ~82.5 ± 3.3%                | ~80 ± 3.2 %                             |
| <sup>4</sup> He                  | $13.6 \pm 0.3\%$                        | $12 \pm 2\%$                   | $15.2 \pm 3.3 \%$           | 19.0 ± 3.2 %                            |
| CH <sub>4</sub>                  | $(1.81 \pm 0.34) \times 10^{-3}$        | $(4.7 \pm 0.2) \times 10^{-3}$ | ~2.3 %                      | ~1-2 %                                  |
| NH <sub>3</sub>                  | $(6.1 \pm 2.8) \times 10^{-4}$          | $(1.6 \pm 1.1) \times 10^{-4}$ | <100 ppb                    | <600 ppb                                |
| H <sub>2</sub> O                 | 520 <sup>+340</sup> <sub>-240</sub> ppm | 2-20 ppb                       |                             |   |
| $H_2S$                           | 67 ± 4 ppm                              | <0.4 ppm                       | <0.8 ppm                    | <3 ppm                                  |
| HD                               | $45 \pm 12 \text{ ppm}$                 | $110 \pm 58 \text{ ppm}$       | ~148 ppm                    | ~192 ppm                                |
| 13CH4                            | $19 \pm 1$ ppm                          | 51±2 ppm                       |                             |   |
| C <sub>2</sub> H <sub>6</sub>    | 5.8 ± 1.5 ppm                           | $7.0 \pm 1.5$ ppm              |                             |   |
| PH <sub>3</sub>                  | $1.1 \pm 0.4$ ppm                       | $4.5 \pm 1.4$ ppm              |                             |   |
| CH <sub>3</sub> D                | $0.20 \pm 0.04 \text{ ppm}$             | $0.30\pm0.02~\text{ppm}$       | ~8.3 ppm                    | ~12 ppm                                 |
| C <sub>2</sub> H <sub>2</sub>    | $0.11 \pm 0.03 \text{ ppm}$             | $0.30\pm0.10\ ppm$             | ~10 ppb                     | 60 <sup>+140</sup> <sub>-40</sub> ppb   |
| HCN                              | $60 \pm 10 \text{ ppb}$                 | <4 ppb                         | <15 ppb                     | $0.3 \pm 0.15$ ppb                      |
| HC <sub>3</sub> N                |   |                                | <0.8 ppb                    | <0.4 ppb                                |
| C <sub>2</sub> H <sub>4</sub>    | $7 \pm 3 \text{ ppb}$                   | ~0.2 ppb <sup>b</sup>          |                             |   |
| CO <sub>2</sub>                  | 5-35 ppb                                | 0.3 ppb                        | 40 ± 5 ppt                  |   |
| C <sub>2</sub> H <sub>6</sub>    |   |                                | $10 \pm 1$ ppb              | 1.5 <sup>+2.5</sup> <sub>-0.5</sub> ppm |
| CH <sub>3</sub> C <sub>2</sub> H | 2.5 <sup>+2</sup> <sub>-1</sub> ppb     | 0.6 ppb                        | $0.25 \pm 0.03 \text{ ppb}$ |   |
| СО                               | $1.6 \pm 0.3$ ppb                       | $1.4 \pm 0.7 \text{ ppb}$      | <40 ppb                     | 0.65 ± 0.35 ppm                         |
| CH <sub>3</sub> CN               |   |                                |                             | <5 ppb                                  |
| GeH <sub>4</sub>                 | 0.7 <sup>+0.4</sup> <sub>-0.2</sub> ppb | $0.4 \pm 0.4$ ppb              |                             |   |
| C <sub>4</sub> H <sub>2</sub>    | $0.3 \pm 0.2 \text{ ppb}$               | 0.09 ppb                       | $0.16 \pm 0.02 \text{ ppb}$ |   |
| AsH <sub>3</sub>                 | $0.22 \pm 0.11$ ppb                     | $2.1 \pm 1.3 \text{ ppb}$      |                             |   |

Lodders, 2010

-assuming a total stratospheric column density of 1.54%10<sup>-2</sup> cm<sup>-2</sup>. From Lodders & Fegley 1998 and updates: Mahaffy et al. 2000, Atreya et al. 2003, Lodders 2004, Wong et al. 2004



Clouds of Jupiter and Saturn



Sanchez-Lavega et al.



## Galileo probe (entry: December 7, 1995)





### Dry atmosphere ?

- Brightness of the sky abruptly drops off at a pressure level of 0.6 bars, indicating an **ammonia cloud layer** above this height. The tenuous cloud layer detected by the NEP was *not* seen by this experiment.
- Clouds are patchy and that the Probe went through a relatively clear area.



- The atmosphere has much less oxygen than the Sun's atmosphere, implying a surprisingly dry atmosphere.
- Planetary scientists had expected oxygen to be enriched relative to the solar value due to impacts by comets and other small bodies over the 4.5 billion years.



### The probe apparently entered a special location



### Atmospheric chemistry on Titan



#### Atmospheric composition of Titan (Coustenis 2007)

| onstituent                                 | Mole Fraction (atm. altitude level)                           |  |
|--|---|--|
| lajor                                      |   |  |
| Molecular nitrogen, N <sub>2</sub>         | 0.98  |  |
| Methane, CH4                               | $4.9\times 10^{-2}(\text{surface})$                           |  |
|  | $1.4-1.6 \times 10^{-2}$ (stratosphere)                       |  |
| Monodeuterated methane, CH <sub>3</sub> D  | $6\times 10^{-6}$ (in CH_3D, in stratosphere.)                |  |
| Argon, <sup>36Ar</sup>                     | $2.8 \times 10^{-7}$  |  |
| 40 <sub>A</sub> r                          | $4.3 \times 10^{-5}$  |  |
| linor                                      |   |  |
| Hydrogen, H <sub>2</sub>                   | ~0.0011   |  |
| Ethane, C <sub>2</sub> H <sub>6</sub>      | $1.5\times10^{-5}(\text{around}\;130\;\text{km})$             |  |
| Propane, C3H8                              | $5\times10^{-7}(\text{around}\;125\text{ km})$                |  |
| Acetylene, C <sub>2</sub> H <sub>2</sub>   | $4\times10^{-6}(\text{around 140 km})$                        |  |
| Ethylene, C2H4                             | $1.5\times10^{-7}(\text{around}\;130\;\text{km})$             |  |
| Methylacetylene, CH3C2H                    | $6.5\times10^{-9}(\text{around 110 km})^{\rm c}$              |  |
| Diacetylene, C <sub>4</sub> H <sub>2</sub> | $1.3\times10^{-9}(\textrm{around}\ 110\ \textrm{km})^{\rm c}$ |  |
| Cyanogen, C <sub>2</sub> N <sub>2</sub>    | $5.5\times10^{-9}(\textrm{around}~120~\textrm{km})^{0}$       |  |
| Hydrogen cyanide, HCN                      | $1.0\times 10^{-7}(\text{around}\;120\;\text{km})^{\rm c}$    |  |
|  | $5\times10^{-7}(around~200~km)^b$                             |  |
|  | $5\times10^{-6}(around~500~km)^{b}$                           |  |
| Cyanoacetylene, HC3N                       | $1\times 10^{-9}(\text{around}\ 120\ \text{km})^{\rm e}$      |  |
|  | $1\times 10^{-7}(\text{around 500 km})^{b}$                   |  |
| Acetonitrile, CHyCN                        | $1\times 10^{-8}(\text{around 200 km})^{c}$                   |  |
|  | $1\times 10^{-7}(\text{around 500 km})$                       |  |
| Water, H <sub>2</sub> O                    | $8\times 10^{-9}~({\rm at}~400~{\rm km})^d$                   |  |
| Carbon monoxide, CO                        | $4 \times 10^{-5}$ (uniform profile) <sup>e</sup>             |  |
| Carbon dioxide, CO2                        | $1.5 \times 10^{-8}$ (around 120 km)                          |  |















Cassini's Visual and Infrared Mapping Spectrometer (VIMS)